

FUNCTIONALLY SUBSTITUTED AZINES.

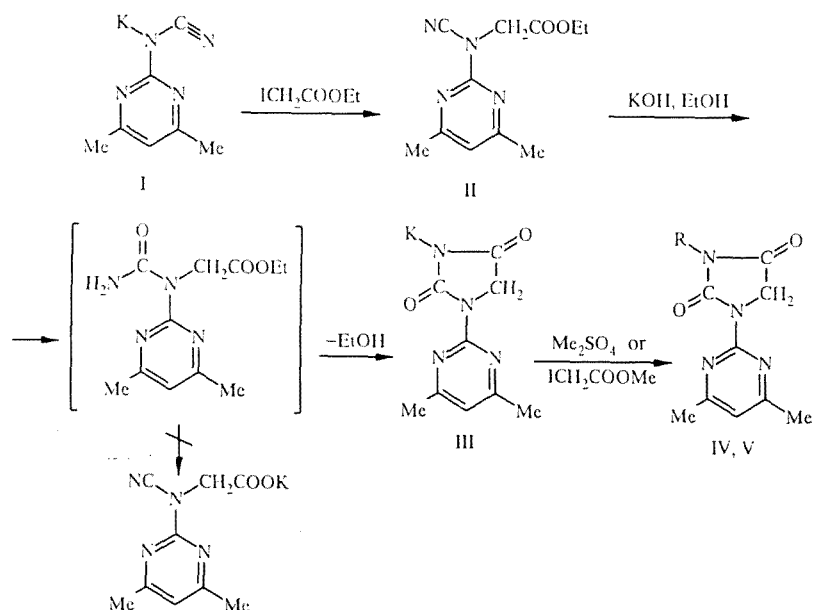
1. SYNTHESIS AND REACTIONS OF 2-(N-CYANO-N-CARBETHOXYMETHYLAMINO)-4,6-DIMETHYLPYRIMIDINE

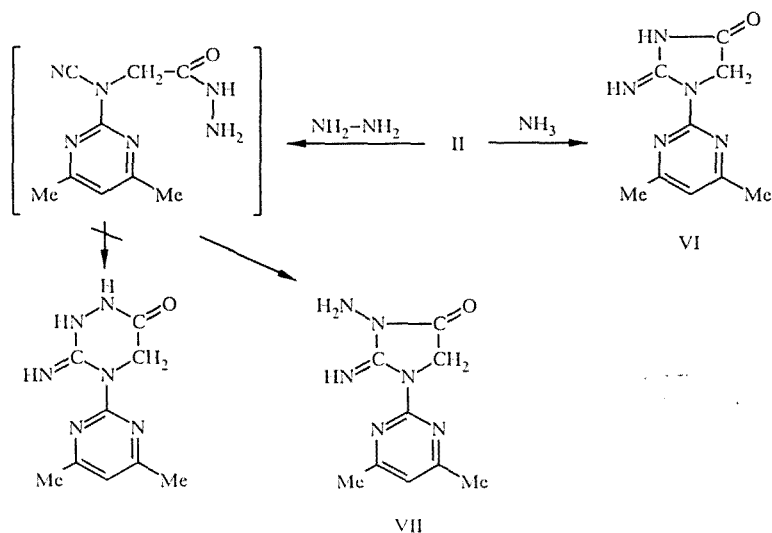
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Treatment of 2-(N-potassium-N-cyanamino)-4,6-dimethylpyrimidine with ethyliodoacetate produces 2-(N-cyano-N-carbethoxymethylamino)-4,6-dimethylpyrimidine. Alkaline hydrolysis of the last gives 2-(3-potassium-2,4-dioxoimidazolidinyl-1)-4,6-dimethylpyrimidine. Ammonolysis and hydrazinolysis give imino- or aminoimidazolidinylpyrimidine, respectively.

Of all possible isomeric cyanaminopyrimidines, only the 2-cyanamino-substituted derivatives are known. These are prepared by the reaction of cyanoguanidine with malonic and acetic esters and β -diketones [1, 2]. The chemistry of the cyanaminopyrimidines is very poorly studied. Thus, only their N-methylation has been described [2]. Moreover, conversions in this series of compounds at the reactive cyanamino group, considering the broad synthetic capabilities of the related cyanamino-1,3,5-triazines [3-8], are promising both in searching for physiologically active compounds and in developing new synthetic approaches for functionally substituted pyrimidines.

Introduction into the cyanamine fragment of these compounds of a carboxyalkyl substituent seemed very interesting. This would open a new synthetic pathway for derivatives of pyrimidinylalkanecarboxylic acids. The only convenient method for preparing these is based on the reaction of amino acid derivatives with chloropyrimidines [9]. Therefore, the available 2-N-potassiumcyanamino-4,6-dimethylpyrimidine (I) was carbethoxyalkylated by ethyliodoacetate in DMF at 70-80 °C in 78% yield to give 2-(N-cyano-N-carbethoxymethylamino)-4,6-dimethylpyrimidine (II). Hydrolysis of II by KOH in alcohol at 20 °C was attempted in order to convert it to the free acid. However, 2-(3-potassium-2,4-dioxoimidazolidinyl-1)-4,6-dimethylpyrimidine (III) was obtained instead of the expected potassium salt of the N-pyrimidinylglycine derivative.





Formation of the intermediate urea derivative probably explains this result. The intermediate eliminates ethanol and cyclizes into III. Reaction of III with dimethylsulfate or methyliodoacetate gives 3-methyl- (IV) or 3-carbomethoxymethyl-(2,4-dioxoimidazolidinyl-1)-4,6-dimethylpyrimidine (V), respectively. Ammonolysis of II over a long period in 25% NH₄OH at 20 °C converts it almost entirely to 2-(2-imino-4-oxoimidazolidinyl-1)-4,6-dimethylpyrimidine (VI). Absorption bands characteristic of the NH₂ group are absent from the IR spectrum of VI. This is consistent with a practically complete shift of the imine—amine tautomeric equilibrium toward the imine form. Under analogous conditions, II easily undergoes hydrazinolysis to form 2-(2-amino-2,4-dioxoimidazolidinyl-1)-4,6-dimethylpyrimidine (VIII).

EXPERIMENTAL

Infrared spectra were recorded on a UR-20 spectrophotometer in mineral oil or KBr pellets. A Varian T-60 (60 MHz) instrument was used to record PMR spectra (TMS internal standard). Mass spectra were obtained on a MX-1303 spectrometer by direct insertion into the source with an ionization energy of 50 eV. Silufol UV-254 plates were used for TLC.

Elemental analyses of the synthesized compounds for C, H, and N agreed with the calculated values.

2-(N-Cyano-N-carbethoxymethyl)amino-4,6-dimethylpyrimidine (II, C₁₁H₁₄N₄O₂). A solution of the ethyl ester of iodoacetic acid (2.14 g, 0.01 mole) in DMF (10 ml) was added dropwise with stirring to I (1.86 g, 0.01 mole) in DMF (3 ml). The mixture was kept at 70–80 °C for 6 h. The DMF was vacuum distilled at 70–80 °C (80 mm Hg). The residue was treated with H₂O (20 ml). Compound II was filtered off. Yield 1.8 g (78%), mp 107–108 °C, R_f 0.45 (heptane–acetone, 1:2). IR spectrum: 1730 (C=O), 2210 cm⁻¹ (C≡N). Found: M⁺ = 234.

2-(3-Potassium-2,4-dioxoimidazolidinyl-1)-4,6-dimethylpyrimidine (III, C₉H₉KN₄O₂). Compound II (2.34 g, 0.01 mole) was added to KOH (84%, 0.7 g, 0.01 mole) in ethanol (20 ml). The mixture was stirred at 20 °C for 6 h. The precipitate of III was filtered off. Yield 1.76 g (73%), mp >270 °C. PMR spectrum (D₂O): 2.42 (6H, s, 2CH₃), 4.42 (2H, s, CH₂), 6.95 ppm (1H, s, CH).

2-(3-Methyl-2,4-dioxoimidazolidinyl-1)-4,6-dimethylpyrimidine (IV, C₁₀H₁₂N₄O₂). Dimethylsulfate (1 ml, 0.01 mole) was added dropwise with stirring at –5 to 0 °C to a suspension of III (2.44 g, 0.01 mole) in acetone (20 ml). The mixture was kept for 3 h at 45–50 °C. The acetone was distilled off. The residue was soaked in H₂O (10 ml). Compound IV was filtered off. Yield 1.2 g (59%), mp 54–55 °C, R_f 0.52 (octane–acetone, 1:2.5). PMR spectrum (acetone-D₆): 2.40 (6H, s, 2CH₃), 3.77 (3H, s, NCH₃), 4.70 (2H, s, CH₂), 6.95 ppm (1H, s, CH).

2-(3-Carbomethoxymethyl-2,4-dioxoimidazolidinyl-1)-4,6-dimethylpyrimidine (V, C₁₂H₁₄N₄O₄). A mixture of methylchloroacetate (1.18 g, 0.01 mmole) and sodium iodide (1.5 g, 0.01 mole) in DMF (10 ml) was kept for 24 h at 20 °C. The reaction mixture was filtered. The filtrate (solution of methyliodoacetate) was added dropwise to a suspension of III (2.44 g, 0.01 mole) in DMF (10 ml). The mixture was kept at 70–80 °C for 4 h. The DMF was distilled off (80 mm Hg). The

residue was treated with H₂O (5 ml). Compound V was filtered off. Yield 1.6 g (67%), mp 58-60°C, R_f 0.47 (heptane–ethylacetate, 4:2). IR spectrum: 1560, 1605 (C=C, C=N), 1700 (C=O ring), 1760 cm⁻¹ (C=O). Found: M⁺ = 248.

2-(2-Imino-4-oxoimidazolidinyl-1)-4,6-dimethylpyrimidine (VI, C₉H₁₁N₅O). A mixture of II (2.34 g, 0.01 mole) and NH₄OH (8-10 ml, 20%) was stirred for 24 h at 20 °C. Water (10 ml) was added. Compound VI was filtered off. Yield 2.0 g (90%), mp > 270 °C. IR spectrum: 1550, 1600 (C=C, C=N), 1660 (C=NH), 1700 (C=O), 3300 cm⁻¹ (NH). PMR spectrum (CDCl₃ + CD₃OD): 2.40 (6H, s, 2CH₃), 4.40 (2H, s, CH₂), 6.77 ppm (1H, s, CH).

2-(3-Amino-2-imino-4-oxoimidazolidinyl-1)-4,6-dimethylpyrimidine (VII, C₉H₁₂N₆O). Hydrazine hydrate (3.4 ml, 55%) was added dropwise to II (2.34 g, 0.01 mole) in H₂O (151 ml). The mixture was kept for 5 h at 20°C. The precipitate of VII was filtered off. Yield 2.1 g (91%), mp > 270°C, R_f 0.40 (hexane–acetone, 1:1). Found: M⁺ = 220. IR spectrum: 1560, 1605 (C=C, C=N), 1690 (C=O ring), 3375 cm⁻¹ (NH₂). PMR spectrum (CDCl₃ + CD₃OD): 2.42 (6H, s, 2CH₃), 4.43 (2H, s, CH₂), 6.75 ppm (1H, s, CH).

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